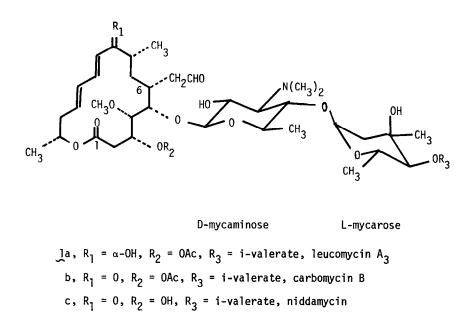
SYNTHESIS OF A FUNCTIONALIZED D-GLUCOSE: A SYNTHON FOR THE CARBOMYCINS (MAGNAMYCINS)

Frederick E. Ziegler*, Paul J. Gilligan and Utpal R. Chakraborty

Department of Chemistry, Yale University New Haven, Connecticut 06520

Abstract: D-Glucose has been converted into its functionalized derivative $\underline{7b}$ a synthem ($C_1 - C_6$) of the carbomycin macrolide antibiotics.

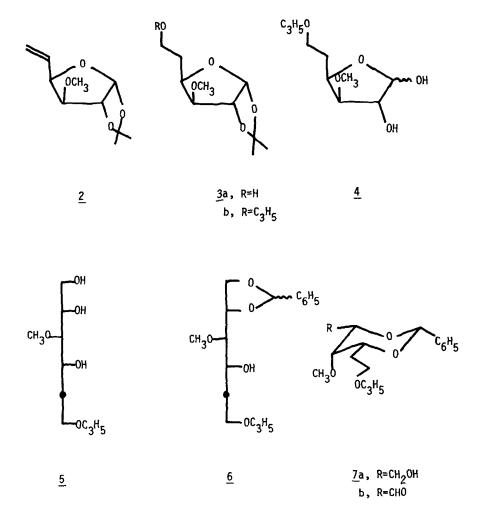
The 16-membered macrolides <u>1</u> are representative of a group of naturally occurring antibiotics which are active against gram positive organisms and certain *mycoplasm* strains.^{1,2} As part of a program to synthesize members of this class, we report in this Letter a functionalized D-glucose which serves as the chiral C_1 - C_6 synthon of the aglycone portion of these antibiotics.³



The olefin <u>2</u> was prepared by a four step sequence (82% yield) from diacetone glucose by modified literature procedures.⁴ Hydroboration of olefin <u>2</u> (disiamylborane, 25°C, 18h; NaOH/H₂O₂, <20°C) provided the alcohol <u>3a</u>⁵ in 88% yield: bp 110°C (0.25 mm); $[\alpha]_D^{31}$ -47.6° (c 1.01, CHC1₃); IR (neat) 3450 cm⁻¹; NMR (CDC1₃, 270 MHz): $\delta 5.88$ (1H, d, J = 4 Hz, C₁-H), 4.59 (1H, d, J = 4 Hz, C₂-H), 4.32 (1H, m, C₄-H), 3.76 (2H, m, -CH₂OH), 3.62 (1H, d, J = 3 Hz, C₃-H), 3.42 (3H, s, -OCH₃), 2.65 (1H, m, OH), 2.05-1.81 (2H, m, CH₂), and 1.50, 1.33 (2 x 3H, s, acetonide). The hydroxyl group was protected by allylation (NaH, THF reflux; allyl bromide, 25°C, 18h) to provide <u>3b</u> in 94% yield: bp 89-90°C (0.15 mm); $[\alpha]_D^{31}$ -41.9° (c 1.00, CHC1₃); NMR (CDC1₃, 270 MHz) $\delta 5.89$ -5.84 (1H, m), 5.87 (1H, d, J = 4 Hz), 5.30-5.14 (2H, m), 4.57 (1H, d, J = 4 Hz), 4.29 (1H, m), 3.98 (1H, d, d, J = 5 Hz, 1 Hz), 3.60-3.53 (4H, m), 3.40 (3H, s), 2.00-1.93 (2H, m) and 1.49, 1.32 (2 x 3H, s). The allyl protecting group could be cleaved with tris-triphenylphosphine rhodium chloride according to the procedure of Corey⁶ to regenerate the alcohol <u>3a</u>.

The acetonide group of <u>3b</u> was removed (4% H_2SO_4 :dioxane, 1:1 (V/V), 100°C, 2h) to afford the anomeric hemiacetals <u>4</u> (positive Fehlings test) as a viscous oil in 92% yield: IR (neat) 3400 cm⁻¹; NMR (CDCl₃, 270 MHz) δ 3.44 and 3.41 (3H, 1:2 ratio, 0CH₃). As anticipated, the hemiacetal was quantitatively reconverted to <u>3b</u> (2,2-dimethoxypropane, p-TsOH, 18h, reflux), precluding the direct formation of the dioxolane <u>7b</u> (as its acetonide) from <u>4</u>. Reduction of <u>4</u> (NaBH₄, C₂H₅OH, 25°C, 60h) provided the crude triol <u>5</u> (negative Fehlings and negative boron tests) in 100% yield: IR (neat) 3400 cm⁻¹.

The triol was transformed $(C_{6}H_{5}CHO, H_{2}SO_{4}, CH_{2}Cl_{2}, 4\text{\AA} molecular sieves, 2 days, 25°C)$ into a 7:1 mixture⁷ (35% yield from <u>3b</u>) of <u>7a</u>: mp 63-64°C; IR (CHCl_{3}) 3550 cm⁻¹; $[\alpha]_{D}^{31}$ +41.3° (c 1.83, CHCl_{3}); NMR (CDCl_{3}, 270 MHz) δ 7.55-7.49 (2H, m), 7.39-7.28 (3H, m), 6.00-5.85 (1H, m), 5.61 (1H, s, benzylidene methine), 5.34-5.15 (2H, m), 4.10-3.94 (6H, m), 3.84-3.78 (1H, m), 3.71-3.57 (2H, m), 3.52 (3H, s), 3.12 (1H, s), 2.21-2.08 (1H, m), and 1.98-1.85 (1H, m) and <u>6</u>: NMR (CDCl_{3}, 270 MHz) δ 5.96 (1H, s) and 5.82 (1H, s) (benzylidene methines). Collins oxidation of alcohol <u>7a</u> provided 5-deoxy-6-0-ally1-2,4-di-0-benzylidene-3-0-methy1-D-glucose <u>7b</u> in 60% yield: mp 79-80°C; $[\alpha]_{D}^{25}$ +63.4° (c 0.95, CHCl_{3}); IR (CHCl_{3}) 2860, 2750 and 1730 cm⁻¹; NMR (CDCl_{3}, 270 MHz) δ 9.76 (1H, s, CHO), 7.60-7.26 (5H, m, pheny1), 5.99-5.89 (1H, m, -CH=), 5.66 (1H, s, benzylidene methine), 5.28-5.19 (2H, m, =CH₂), 4.32 (1H, d, J = 2 Hz, C_{2} -<u>H</u>), 4.12-4.05 (1H, m, C_{4} -H), 3.99 (2H, d, J = 6 Hz), 3.72-3.45 (2H, m, -CH₂O-),



3.54 (1H, br.s, CH-OCH₃), 3.43 (3H, s, -OCH₃), 2.19-2.07 (1H, m), and 1.98-1.86 (1H, m).

(+) Deoxyglucose (<u>7b</u>) displays the correct absolute stereochemistry to serve as a chiral precursor of C₁-C₆ of the macrolide aglycones. Synthon <u>7b</u> will be utilized by further functionalization of the aldehyde carbon.

ACKNOWLEDGMENTS: This research was supported by the NIH (AI-15617-01), Hoffmann-LaRoche (Nutley), and NIH Research Grant No. 1-P07-PR00798 (Bruker HX-270) from the Division of Research Sources.

REFERENCES:

- 1. K. Tatsua, A. Tanaka, K. Fujimoto, and M. Kinoshita, J. Am. Chem. Soc., 99, 5826 (1977).
- 2. S. Omura and A. Nakagawa, J. Antibiot., 28, 401 (1975).
- 3. We thank Professor K.C. Nicolaou, University of Pennsylvania, for a preprint of his work in this area.
- 4. J.S. Josan and F.W. Eastwood, Carbohyd. Res., 7, 161 (1968).
- All new distilled and crystalline compounds gave correct combustion analyses.
- 6. E.J. Corey and J.W. Suggs, J. Org. Chem., <u>38</u>, 3224 (1973).
- Separated by flash chromatography, ethyl acetate-hexanes (3/1). W.C. Still, M. Kahn and A. Mitra, J. Org. Chem., <u>43</u>, 2923 (1978).

(Received in USA 30 May 1979)